



*THE ALBERTA LAKE MANAGEMENT SOCIETY  
VOLUNTEER LAKE MONITORING PROGRAM*

## **2015 Isle Lake Report**

*LAKEWATCH IS MADE POSSIBLE WITH SUPPORT FROM:*



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## **Alberta Lake Management Society's LakeWatch Program**

LakeWatch has several important objectives, one of which is to collect and interpret water quality data on Alberta Lakes. Equally important is educating lake users about their aquatic environment, encouraging public involvement in lake management, and facilitating cooperation and partnerships between government, industry, the scientific community and lake users. LakeWatch Reports are designed to summarize basic lake data in understandable terms for a lay audience and are not meant to be a complete synopsis of information about specific lakes. Additional information is available for many lakes that have been included in LakeWatch and readers requiring more information are encouraged to seek those sources.

ALMS would like to thank all who express interest in Alberta's aquatic environments and particularly those who have participated in the LakeWatch program. These people prove that ecological apathy can be overcome and give us hope that our water resources will not be the limiting factor in the health of our environment.

This report was published before the completion of the data validation process.

### **Acknowledgements**

The LakeWatch program is made possible through the dedication of its volunteers. We would like to thank Carla & Adam Laidlaw for their assistance with sampling Gull Lake in 2015. We would also like to thank Laticia McDonald, Ageleky Bouzetos, and Mohamad Youssef who were summer technicians with ALMS in 2015. Executive Director Bradley Peter was instrumental in planning and organizing the field program. Mike Bilyk was involved in the training aspects of the program. Lisa Reinbolt was responsible for data management. This report was prepared by Bradley Peter and Alicia Kennedy. The Beaver River Watershed Alliance (BRWA), the Lakeland Industry and Community Association (LICA), the Alberta Environmental Monitoring Evaluation and Reporting Agency (AEMERA), and Environment Canada, were major sponsors of the program.

## ISLE LAKE:

Isle Lake is located in the counties of Lac Ste. Anne and Parkland, 80 km west of the City of Edmonton. The Hamlet of Gainford, established in 1942, is situated on the southwest shore. In 1879, the Hudson Bay Company set up a trading post at nearby Lac Ste. Anne, and settlers began arriving in 1905 as agricultural lands became available.<sup>1</sup> Today, several subdivisions are registered along the shoreline and the lake is heavily used for recreation. Sport fishing is popular, and species include northern pike (*Esox lucius*), yellow perch (*Perca flavescens*), burbot (*Lota lota*), white suckers (*Catostomus commersoni*), and walleye (*Sanders vitreus*).<sup>1</sup>

Isle Lake is long and moderately shallow (Figure 1; maximum depth 7.5 m). Isle Lake has experienced many stressors, including high phosphorus content which results in cyanobacteria blooms, and a population of the invasive plant Flowering Rush which has the potential to reproduce and spread rapidly. The Alberta Conservation Association is currently working on a Lake Isle Fisheries Restoration Project in an effort to restore a healthy recreational fishery and reduce phosphorus loading from the watershed.

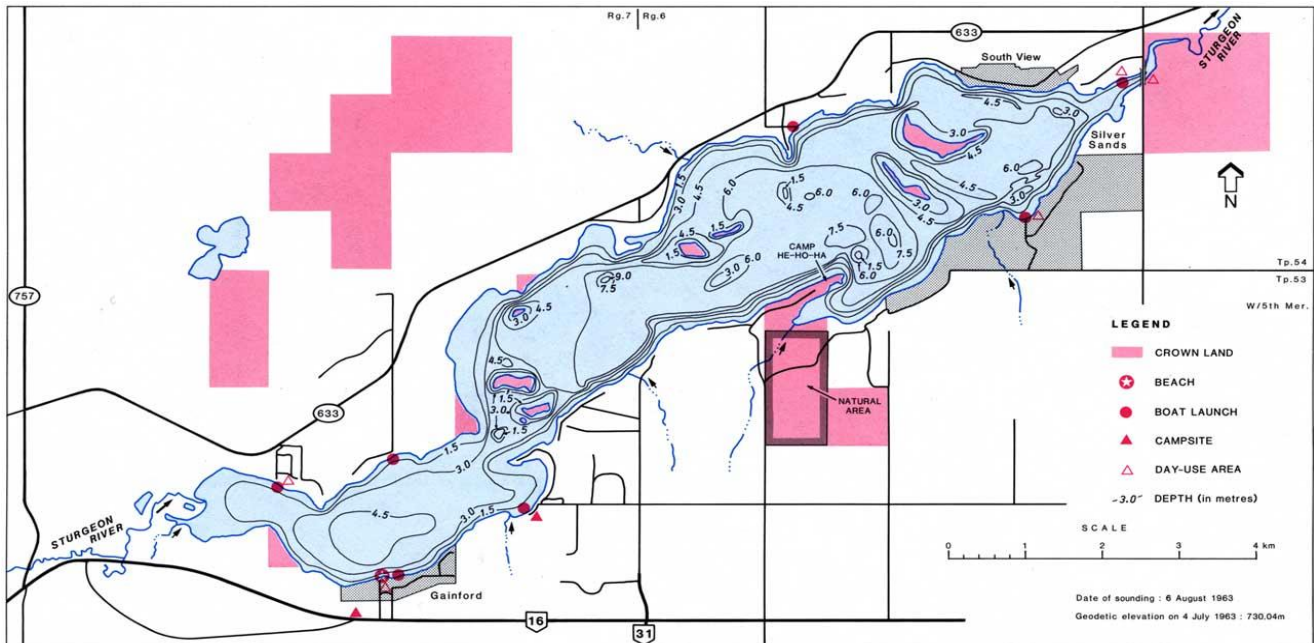


Figure 1 – Bathymetry and shoreline features of Isle Lake.<sup>1</sup>

<sup>1</sup> Mitchell, P. and E. Prepas. 1990. Atlas of Alberta Lakes, University of Alberta Press. Retrieved from <http://sunsite.ualberta.ca/projects/alberta-lakes/>

## WATER LEVELS:

*There are many factors influencing water quantity. Some of these factors include the size of the lakes drainage basin, precipitation, evaporation, water consumption, ground water influences, and the efficiency of the outlet channel structure at removing water from the lake.*

Water levels at Isle Lake have changed little since monitoring began in 1972 (data available until 2012; Figure 2). Between the historical maximum of 730.5 meters above sea level (m asl) in 1989 and the historical minimum of 729.2 m asl in 2010, water levels have only changed by 1.3 m (Figure 2). With a drainage basin area approximately 11 times the size of the lake, Isle Lake receives large amounts of input in the form of runoff. The Sturgeon River runs into Isle Lake from the West and exits the lake to the East, which then flows into neighbouring Lac St. Anne. Environment Canada data was available up to 2013 when Isle Lake measured an average of 729.8 m asl.

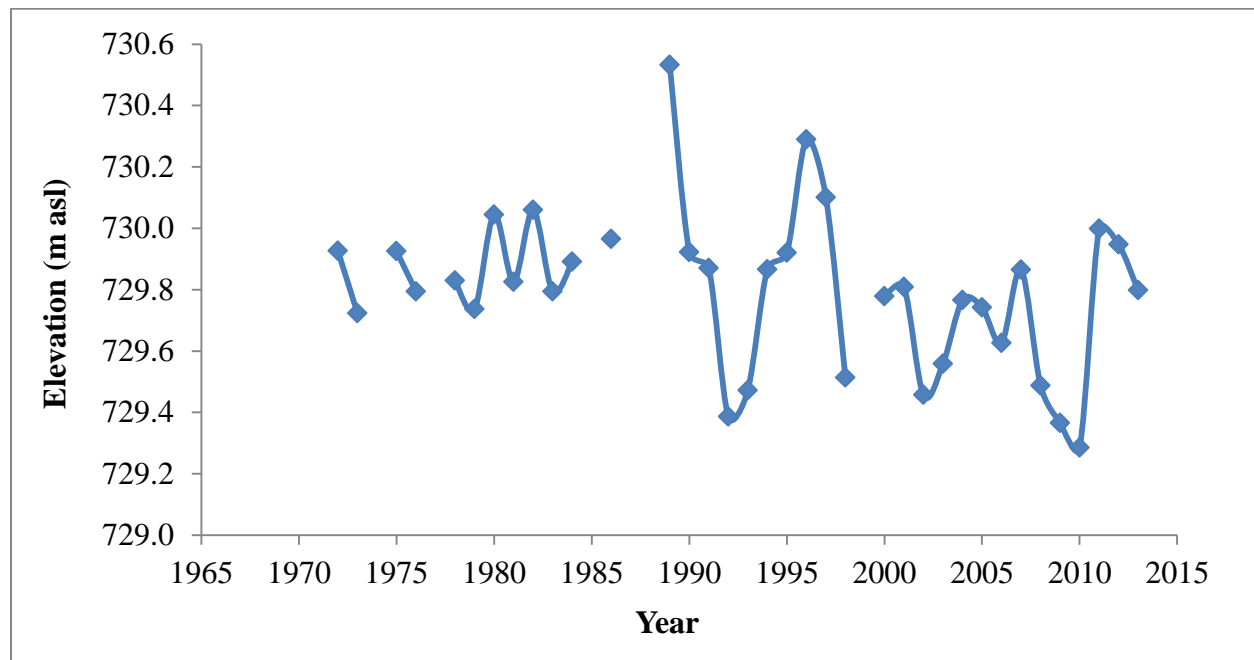


Figure 2 – Water levels measured by Environment Canada for Lake Isle in meters above sea level (m asl).

## WATER CLARITY AND SECCHI DEPTH:

*Water clarity is influenced by suspended materials, both living and dead, as well as dissolved colored compounds in the water column. During the melting of snow and ice in spring, lake water can become turbid (cloudy) from silt transported into the lake. Lake water usually clears in late spring but then becomes more turbid with increased algal growth as the summer progresses. The easiest and most widely used measure of lake water clarity is the Secchi disk depth.*

Water clarity was poor at Isle Lake in 2015 (Figure 3). Cyanobacteria blooms had an obvious impact on water clarity, as the Secchi depth decreased from a maximum of 3.75 m on June 4<sup>th</sup> to only 1.00 m on July 4<sup>th</sup>. Secchi disk depth remained low for the rest of the summer, measuring a minimum of 0.75 m on both August 15<sup>th</sup> and September 26<sup>th</sup>. On average, Secchi disk depth measured 1.45 m in 2015 – this falls on the low end of the historical variation observed at Isle Lake (Table 2).

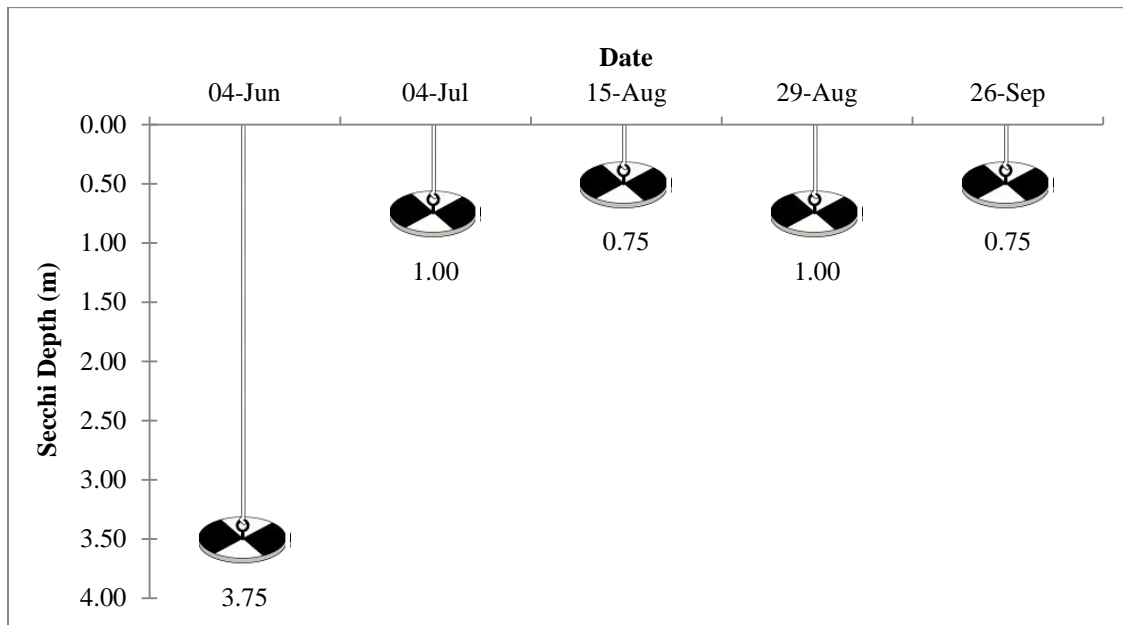


Figure 3 – Secchi disk depth in meters measured five times over the course of the summer at Isle Lake.

#### **WATER TEMPERATURE AND DISSOLVED OXYGEN:**

*Water temperature and dissolved oxygen profiles in the water column can provide information on water quality and fish habitat. The depth of the thermocline is important in determining the depth to which dissolved oxygen from the surface can be mixed. Please refer to the end of this report for descriptions of technical terms.*

Water temperature in 2015 showed weak thermal stratification on both June 6<sup>th</sup> and July 4<sup>th</sup> (Figure 4a). After July 4<sup>th</sup>, no thermal stratification was observed, and the water column remained relatively isothermal. A maximum surface water temperature of 21.40 °C was observed on July 4<sup>th</sup> and a maximum bottom temperature of 20.44 °C was observed on August 15<sup>th</sup>. Temperature in the water column has important implications for fish health – a water column which is warm from top to bottom provides little refuge for temperature-sensitive fish, though is common in shallow lakes across Alberta.

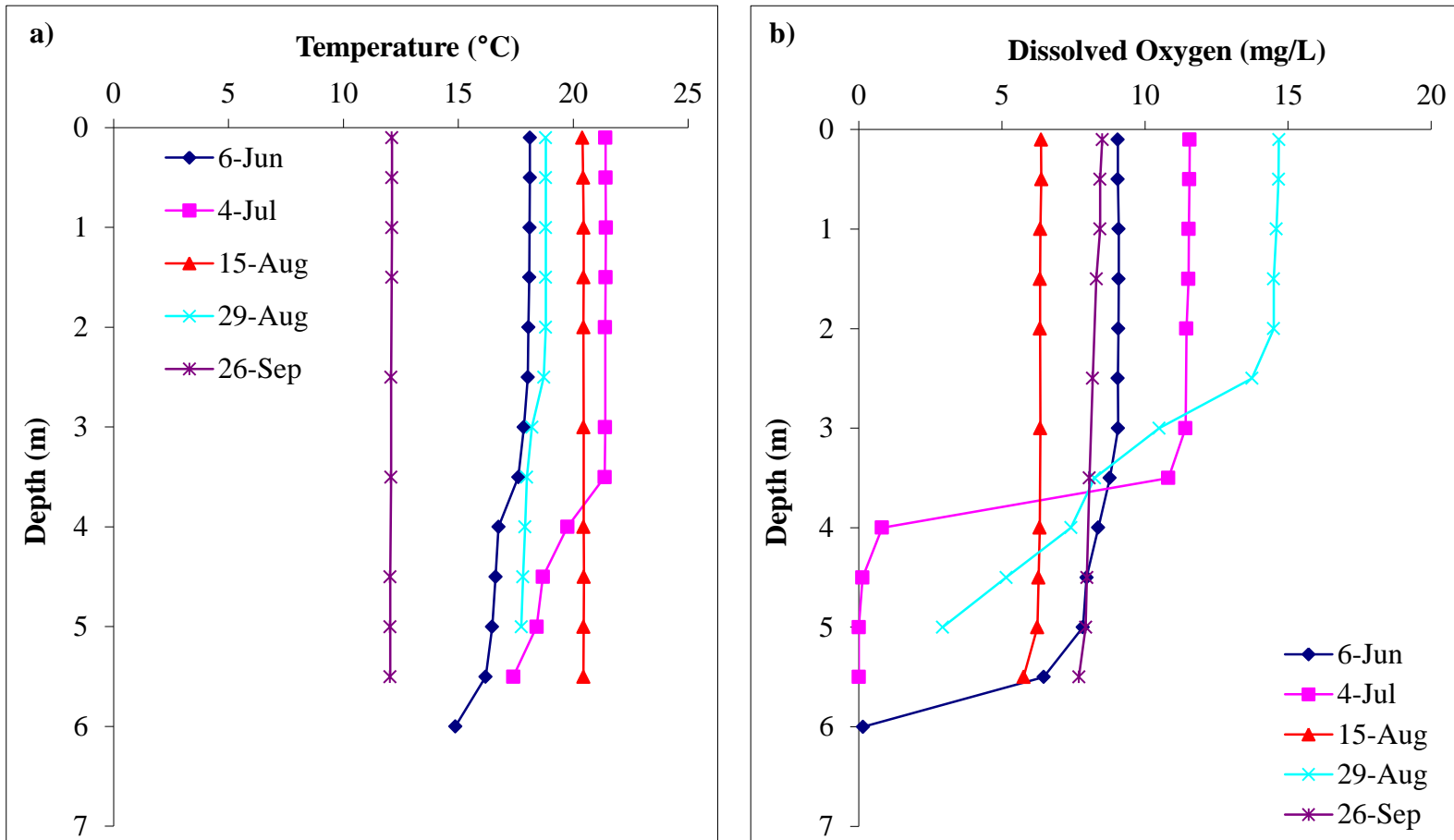


Figure 4 – Temperature (°C) and dissolved oxygen (mg/L) measured five times over the course of the summer at Isle Lake.

Oxygen in Isle Lake varied greatly over the course of the summer (Figure 4b). In the spring, Isle Lake was well oxygenated, with the entire water column falling above the Canadian Council for Ministers of the Environment (CCME) guideline for the Protection of Aquatic Life (PAL) of 6.5 mg/L. On July 4<sup>th</sup>, a large cyanobacteria bloom (Figure 5) resulted in elevated levels of dissolved oxygen at Isle Lake’s surface (11.56 mg/L). However, due to thermal stratification, anoxia was observed by 4.0 m. After the collapse of this bloom, oxygen concentrations dropped, and the entire water column fell below the CCME PAL guidelines of 6.5 mg/L on August 15<sup>th</sup>. On August 29<sup>th</sup>, another large cyanobacteria bloom was observed which resulted in greatly elevated dissolved oxygen at the surface of the lake (14.68 mg/L). Finally, on September 26<sup>th</sup>, when the water column was well mixed, the lake appeared well oxygenated.

**WATER CHEMISTRY:**

*ALMS measures a suite of water chemistry parameters. Phosphorus, nitrogen, and chlorophyll-a are important because they are indicators of eutrophication, or excess nutrients, which can lead to harmful algal/cyanobacteria blooms. One direct measure of harmful cyanobacteria blooms*

are *Microcystins*, a common group of toxins produced by cyanobacteria. See Table 1 for a complete list of parameters.

Average total phosphorus measured 0.19 mg/L in 2015. This value falls into the hypereutrophic, or extremely productive, classification, and falls well within the historical variation previously observed at Isle Lake (Table 1). Throughout the summer, TP concentration ranged from a minimum of 0.044 mg/L on June 4<sup>th</sup> and a maximum of 0.31 mg/L on August 15<sup>th</sup> (Figure 5). Phosphorus is often the primary nutrient driving cyanobacteria blooms in Alberta lakes. Values in Figure 5 have been reported in µg/L for visualization purposes.

Average chlorophyll-*a* concentration measured 74 µg/L in 2015. This average falls into the hypereutrophic, or extremely productive, classification, and is the third highest average recorded at Isle Lake (Table 1). Multiple blooms occurred during 2015, however a maximum chlorophyll-*a* concentration of 118 µg/L was observed on August 29<sup>th</sup> (Figure 5).

Finally, total Kjeldahl nitrogen (TKN) measured an average of 2.3 mg/L in 2015. This average falls into the hypereutrophic, or extremely productive, classification, and is the third highest average recorded at Isle Lake (Table 1). TKN increased gradually throughout the summer, measuring a maximum of 2.9 mg/L.

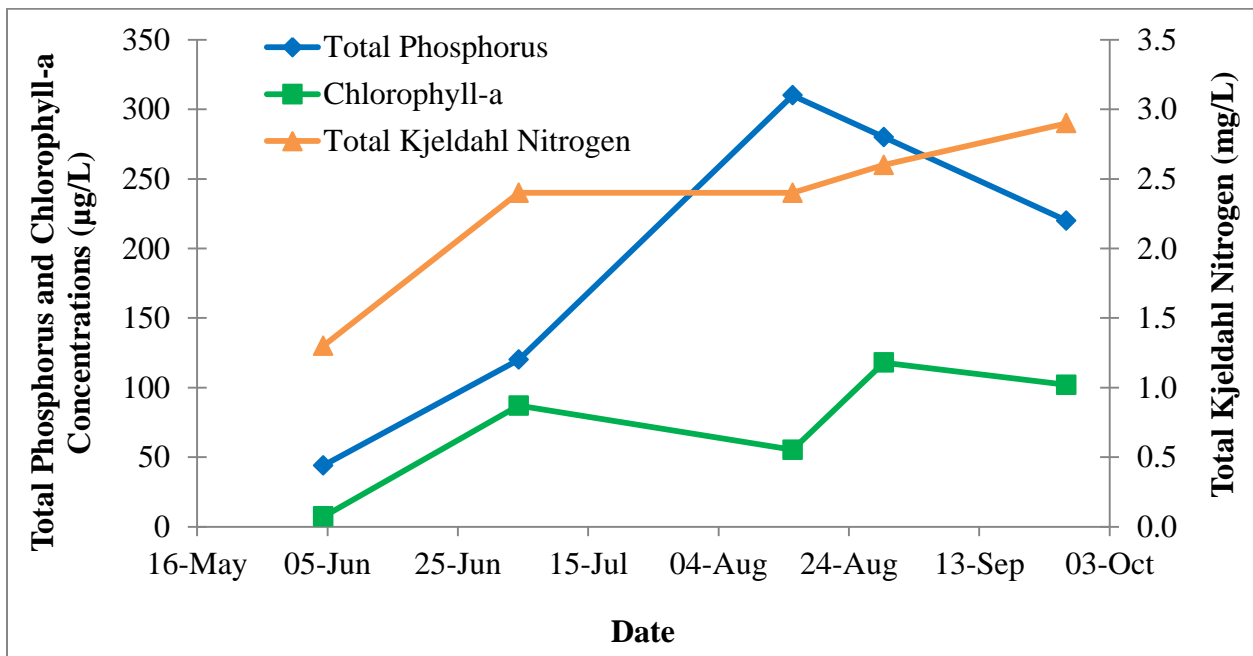


Figure 5 – Dissolved oxygen (mg/L) and temperature (°C) measured five times over the course of the summer at Isle Lake.

Average pH measured 8.88, which is well above neutral. Similar to many other lakes in Alberta, Isle Lake has high bicarbonate concentration (170 mg/L HCO<sub>3</sub>) and alkalinity (162 mg/L CaCO<sub>3</sub>) which help to buffer the lake against changes to pH (Table 2). Concentrations of ions in

the lake contribute to a low conductivity of 354 uS/cm. Dominant ions in Isle Lake include sodium (34 mg/L), calcium (25 mg/L), and chloride (12 mg/L).

Metals were measured three times over the summer at Isle Lake – all concentrations fell within their respective guidelines (Table 3)

#### **MICROCYSTIN:**

*Microcystins are toxins produced by cyanobacteria (blue-green algae) which, when ingested, can cause severe liver damage. Microcystins are produced by many species of cyanobacteria which are common to Alberta's Lakes, and are thought to be the one of the most common cyanobacteria toxins. In Alberta, recreational guidelines for microcystin are set at 20 µg/L.*

Microcystin concentrations varied greatly throughout the summer. On August 15<sup>th</sup>, a maximum of 14.08 µg/L was observed, while a minimum of 0.12 µg/L was observed on July 4<sup>th</sup>. As this data represents water composited from across the lake as opposed to a grab sample at a beach, it is important to use caution when recreating in or around cyanobacteria.

Table 1 – Microcystin concentrations measured in Isle Lake in 2015.

<b>Date</b>	<b>Microcystin Concentration (µg/L)</b>
04-Jun-15	0.19
04-Jul-15	0.12
15-Aug-15	14.08
29-Aug-15	11.59
26-Sep-15	5.85
<b>Average</b>	<b>6.366</b>

#### **INVASIVE SPECIES:**

*Quagga and Zebra mussels are invasive species which, if introduced to our lakes, will have significant negative ecological, economical, and recreational impacts. ALMS collects water samples which are analyzed for mussel veligers (juveniles) and monitors substrates for adult mussels. In order to prevent the spread of invasive mussels, always clean, drain, and dry your boat between lakes. To report mussel sightings or mussel-fouled boats, call the confidential Alberta hotline at 1-855-336-BOAT.*

No zebra or quagga mussels were detected in Isle Lake in 2015.



Table 2 - Average Secchi disk depth and water chemistry values for Gull Lake. Previous years averages are provided for comparison.

Parameter	1983	1984	1996	1997	1998	1996-1998	2002	2011	2012	2014	2015
TP (mg/L)	0.14	0.17	0.09	0.18	0.37	/	0.15	0.23	0.25	0.25	0.19
TDP (µg/L)	72	115	37	116	285	/	95	85	124.6	172.6	122
Chl- <i>a</i> (µg/L)	64.7	47.5	48.6	65.6	67.1	/	20	113	117.83	45.38	74
Secchi depth (m)	1.4	1.3	3.1	1.7	2.1	/	2.1	0.66	1.463	1.49	1.45
TKN (mg/L)	1.8	1.8	1.3	1.7	2.1	/	1.6	2.9	2.9	2.2	2.3
NO <sub>2</sub> and NO <sub>3</sub> (µg/L)	/	/	/	/	/	/	25	24.5	0.01	36	3.8
NH <sub>3</sub> (µg/L)	/	/	/	/	/	/	52	78.2	81.45	207.4	80
DOC (mg/L)	/	/	/	/	/	/	/	17.9	/	19.7	19
Ca (mg/L)	/	/	/	/	/	30.6	30	25.3	28.05	27.43	25
Mg (mg/L)	/	/	/	/	/	8.6	10	10.8	10.21	9.319	11
Na (mg/L)	/	/	/	/	/	19	26	32.8	31.36	34.63	34
K (mg/L)	/	/	/	/	/	7	9	9.3	9.49	10.23	11
SO <sub>4</sub> <sup>2-</sup> (mg/L)	/	/	/	/	/	30.6	8.9	4	10.34	6.433	7.5
Cl <sup>-</sup> (mg/L)	/	/	/	/	/	5	6.6	10.3	10.05	10.97	12
CO <sub>3</sub> (mg/L)	/	/	/	/	/	7.6	6.2	3.2	14.48	16.05	14
HCO <sub>3</sub> (mg/L)	/	/	/	/	/	165	189	203	177.75	186.2	170
pH	/	/	/	/	/	7.7-9.0	/	8.32	8.81	8.65	8.88
Conductivity (µS/cm)	/	/	/	/	/	308	/	364	365.25	370	354
Hardness (mg/L)	/	/	/	/	/	112	/	108	112.15	106.67	108
TDS (mg/L)	/	/	/	/	/	166	/	196	202	225.33	202
Microcystin (µg/L)	/	/	/	/	/	/	/	0.96	3.255	2.154	6.37
Total Alkalinity (mg/L CaCO <sub>3</sub> )	/	/	/	/	/	148	153	171	170.5	153.04	162

Note: TP = total phosphorous, TDP = total dissolved phosphorous, Chl-*a* = chlorophyll-*a*, TKN = total Kjeldahl nitrogen. NO<sub>2+3</sub> = nitrate+nitrite, NH<sub>3</sub> = ammonia, Ca = calcium, Mg = magnesium, Na = sodium, K = potassium, SO<sub>4</sub> = sulphate, Cl = chloride, CO<sub>3</sub> = carbonate, HCO<sub>3</sub> = bicarbonate, TDS = total dissolved solid. A forward slash (/) indicates an absence of data.

Table 3 - Average concentrations of metals measured in Isle Lake on June 4<sup>th</sup>, August 15<sup>th</sup>, and September 26<sup>th</sup>. The CCME heavy metal Guidelines for the Protection of Freshwater Aquatic Life (unless otherwise indicated) are presented for reference.

<b>Metals (Total Recoverable)</b>	<b>2012</b>	<b>2014</b>	<b>2015</b>	<b>Guidelines</b>
Aluminum µg/L	19.7	10.65	41.83	100 <sup>a</sup>
Antimony µg/L	0.06525	0.058	0.058	6 <sup>e</sup>
Arsenic µg/L	1.765	1.995	1.650	5
Barium µg/L	78.2	83.6	72.2	1000 <sup>e</sup>
Beryllium µg/L	0.0078	0.004	0.004	100 <sup>d,f</sup>
Bismuth µg/L	0.00665	0.0005	0.0015	/
Boron µg/L	57.45	51.4	64.27	5000 <sup>ef</sup>
Cadmium µg/L	0.0047	0.0025	0.0013	0.085 <sup>b</sup>
Chromium µg/L	0.253	0.385	0.313	/
Cobalt µg/L	0.04645	0.0135	0.0353	1000 <sup>f</sup>
Copper µg/L	0.318	0.44	0.28	4 <sup>c</sup>
Iron µg/L	36.45	40.1	68.2	300
Lead µg/L	0.05275	0.0695	0.0347	7 <sup>c</sup>
Lithium µg/L	16.7	14.4	18.2	2500 <sup>g</sup>
Manganese µg/L	64.5	153.25	128	200 <sup>g</sup>
Mercury @ Surface ng/L	/	/	0.54	26
Mercury @ Bottom ng/L	/	/	0.90	26
Molybdenum µg/L	0.5765	0.333	0.316	73 <sup>d</sup>
Nickel µg/L	0.1094	0.004	0.2450	150 <sup>c</sup>
Selenium µg/L	0.078	0.23	0.067	1
Silver µg/L	0.0043	0.001	0.001	0.1
Strontium µg/L	182	210	190	/
Thallium µg/L	0.0026	0.001425	0.000983	0.8
Thorium µg/L	0.024775	0.012125	0.003417	/
Tin µg/L	0.051	0.0135	0.0207	/
Titanium µg/L	1.475	2.565	2.147	/
Uranium µg/L	0.361	0.2475	0.2637	100 <sup>e</sup>
Vanadium µg/L	0.5145	0.36	0.39	100 <sup>f,g</sup>
Zinc µg/L	1.6215	0.5	0.3833	30

Values represent means of total recoverable metal concentrations.

<sup>a</sup> Based on pH ≥ 6.5; calcium ion concentrations [Ca<sup>+2</sup>] ≥ 4 mg/L; and dissolved organic carbon concentration [DOC] ≥ 2 mg/L.

<sup>b</sup> Based on water Hardness of 300 mg/L (as CaCO<sub>3</sub>)

<sup>c</sup> Based on water hardness > 180mg/L (as CaCO<sub>3</sub>)

<sup>d</sup> CCME interim value.

<sup>e</sup> Based on Canadian Drinking Water Quality guideline values.

<sup>f</sup> Based on CCME Guidelines for Agricultural use (Livestock Watering).

<sup>g</sup> Based on CCME Guidelines for Agricultural Use (Irrigation).

A forward slash (/) indicates an absence of data or guidelines.

# A BRIEF INTRODUCTION TO LIMNOLOGY

## INDICATORS OF WATER QUALITY:

Water samples are collected in LakeWatch to determine the chemical characteristics that characterize general water quality. Though not all encompassing, the variables collected in LakeWatch are sensitive to human activities in watersheds that can cause degraded water quality. For example, nutrients such as phosphorus and nitrogen are important determinants of lake productivity. The concentrations of these nutrients in a lake are impacted (typically elevated) by land use changes such as increased crop production or livestock grazing. Elevated nutrient concentrations can cause increases in undesirable algae blooms resulting in low dissolved oxygen concentrations, degraded habitat for fish and noxious smells. A large increase in nutrients over time may also indicate sewage inputs which in turn may result in other human health concerns associated with bacteria or the protozoan *Cryptosporidium*.

## TEMPERATURE AND MIXING:

Water temperature in a lake dictates the behavior of many chemical parameters responsible for water quality. Heat is transferred to a lake at its surface and slowly moves downward depending on water circulation in the lake. Lakes with a large surface area or a small volume tend to have greater mixing due to wind. In deeper lakes, circulation is not strong enough to move warm water to depths typically greater than 4 or 5 m and as a result cooler denser water remains at the bottom of the lake.

As the difference in temperature between warm surface and cold deeper water increases, two distinct layers are formed. Limnologists call these layers of water the *epilimnion* at the surface and the *hypolimnion* at the bottom. The layers are separated by a transition layer known as the *metalimnion* which contains the effective wall separating top and bottom waters called a *thermocline*. A thermocline typically occurs when water temperature changes by more than one degree within one meter depth. The hypolimnion and epilimnion do not mix, nor do elements such as oxygen supplied at the surface move downward into the hypolimnion. In the fall, surface waters begin to cool and eventually reach the same temperature as hypolimnetic water. At this point the water mixes from top to bottom in what is often called a *turnover* event. Surface water cools further as ice forms and again a thermocline develops this time with 4° C water at the bottom and near 0° C water on the top.

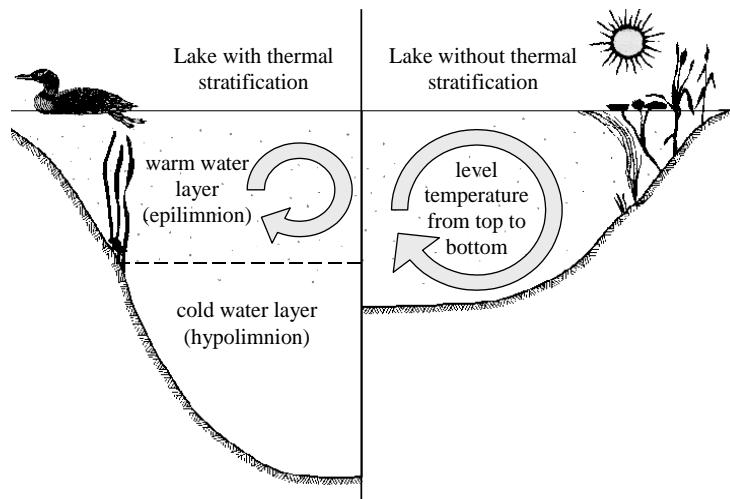


Figure A: Difference in the circulation of the water column depending on thermal stratification.

In spring another turnover event occurs when surface waters warm to 4° C. Lakes with this mixing pattern of two stratification periods and two turnover events are called *dimictic* lakes. In shallower lakes, the water column may mix from top to bottom most of the ice-free season with occasional stratification during periods of calm warm conditions. Lakes that mix frequently are termed *polymictic* lakes. In our cold climate, many shallow lakes are *cold monomictic* meaning a thermocline develops every winter, there is one turnover event in spring but the remainder of the ice free season the lake is polymictic.

#### **DISSOLVED OXYGEN:**

Oxygen enters a lake at the lake surface and throughout the water column when produced by photosynthesizing plants, including algae, in the lake. Oxygen is consumed within the lake by respiration of living organisms and decomposition of organic material in the lake sediments. In lakes that stratify (see temperature above), oxygen that dissolves into the lake at the surface cannot mix downward into the hypolimnion. At the same time oxygen is depleted in the hypolimnion by decomposition. The result is that the hypolimnion of a lake can become **anoxic**, meaning it contains little or no dissolved oxygen. When a lake is frozen, the entire water column can become anoxic because the surface is sealed off from the atmosphere. Winter anoxic conditions can result in a fish-kill which is particularly common during harsh winters with extended ice-cover. Alberta Surface Water Quality Guidelines suggest dissolved oxygen concentrations (in the epilimnion) must not decline below 5 mg•L<sup>-1</sup> and should not average less than 6.5 mg•L<sup>-1</sup> over a seven-day period. However, the guidelines also require that dissolved oxygen concentrations remain above 9.5 mg•L<sup>-1</sup> in areas where early life stages of aquatic biota, particularly fish, are present.

#### **GENERAL WATER CHEMISTRY:**

Water in lakes always contains substances that have been transported by rain and snow or have entered the lake in groundwater and inflow streams. These substances may be dissolved in the water or suspended as particles. Some of these substances are familiar minerals, such as sodium and chloride, which when combined form table salt, but when dissolved in water separate into the two electrically charged components called **ions**. Most dissolved substances in water are in ionic forms and are held in solution due to the polar nature of the water molecule. **Hydrophobic** (water-fearing) compounds such as oils contain little or no ionic character, are non-polar and for this reason do not readily dissolve in water. Although hydrophobic compounds do not readily dissolve, they can still be transported to lakes by flowing water. Within individual lakes, ion concentrations vary from year to year depending on the amount and mineral content of the water entering the lake. This mineral content can be influenced by the amount of precipitation and other climate variables as well as human activities such as fertilizer and road salt application.

#### **PHOSPHORUS AND NITROGEN:**

Phosphorus and nitrogen are important nutrients limiting the growth of algae in Alberta lakes. While nitrogen usually limits agricultural plants, phosphorus is usually in shortest supply in lakes. Even a slight increase of phosphorus in a lake can, given the right conditions, promote algal blooms causing the water to turn green in the summer and impair recreational uses. When

pollution originating from livestock manure and human sewage enters lakes not only are the concentrations of phosphorus and nitrogen increased but nitrogen can become a limiting nutrient which is thought to cause blooms of toxic algae belonging to the cyanobacteria. Not all cyanobacteria are toxic, however, the blooms can form decomposing mats that smell and impair dissolved oxygen concentrations in the lake.

#### **CHLOROPHYLL-A:**

Chlorophyll *a* is a photosynthetic pigment that green plants, including algae, possess enabling them to convert the sun's energy to living material. Chlorophyll *a* can be easily extracted from algae in the laboratory. Consequently, chlorophyll *a* is a good estimate of the amount of algae in the water. Some highly productive lakes are dominated by larger aquatic plants rather than suspended algae. In these lakes, chlorophyll *a* and nutrient values taken from water samples do not include productivity from large aquatic plants. The result, in lakes like Chestermere which are dominated by larger plants known as macrophytes, can be a lower trophic state than if macrophyte biomass was included. Unfortunately, the productivity and nutrient cycling contributions of macrophytes are difficult to sample accurately and are therefore not typically included in trophic state indices.

#### **SECCHI DISK TRANSPARENCY :**

Lakes that are clear are more attractive for recreation, whereas those that are turbid or murky are considered by lake users to have poor water quality. A measure of the transparency or clarity of the water is performed with a Secchi disk with an alternating black and white pattern. To measure the clarity of the water, the Secchi disk is lowered down into the water column and the depth where the disk disappears is recorded. The Secchi depth in lakes with a lot of algal growth will be small while the Secchi depth in lakes with little algal growth can be very deep. However, low Secchi depths are not caused by algal growth alone. High concentrations of suspended sediments, particularly fine clays or glacial till, are common in plains or mountain reservoirs of Alberta. Mountain reservoirs may have exceedingly low Secchi depths despite low algal growth and nutrient concentrations.

The euphotic zone or the maximum depth that light can penetrate into the water column for actively growing plants is calculated as twice the Secchi depth. Murky waters, with shallow Secchi depths, can prevent aquatic plants from growing on the lake bottom. Conversely, aquatic plants can ensure lakes have clear water by reducing shoreline erosion and stabilizing lake bottom sediments. In Alberta, many lakes are shallow and bottom sediments contain high concentrations of nutrients. Without aquatic plants, water quality may decline in these lakes due to murky, sediment laden water and excessive algal blooms. Maintaining aquatic plants in certain areas of a lake is often essential for ensuring good water clarity and a healthy lake as many organisms, like aquatic invertebrates and insects, depend on aquatic plants for food and shelter.

**TROPHIC STATE:**

Trophic state is classification of lakes into four categories of fertility and is a useful index for rating and comparing lakes. From low to high nutrient and algal biomass (as chlorophyll) concentrations, the trophic states are; **oligotrophic, mesotrophic, eutrophic** and **hypereutrophic** (Table 2).

A majority of lakes in Alberta contain naturally high levels of chlorophyll *a* (8 to 25 µg/L) due to our deep fertile soils. These lakes are usually considered fertile and are termed eutrophic. The nutrient and algal biomass concentrations that define these categories are shown in the following table, a figure of Alberta lakes compared by trophic state can be found on the ALMS website.

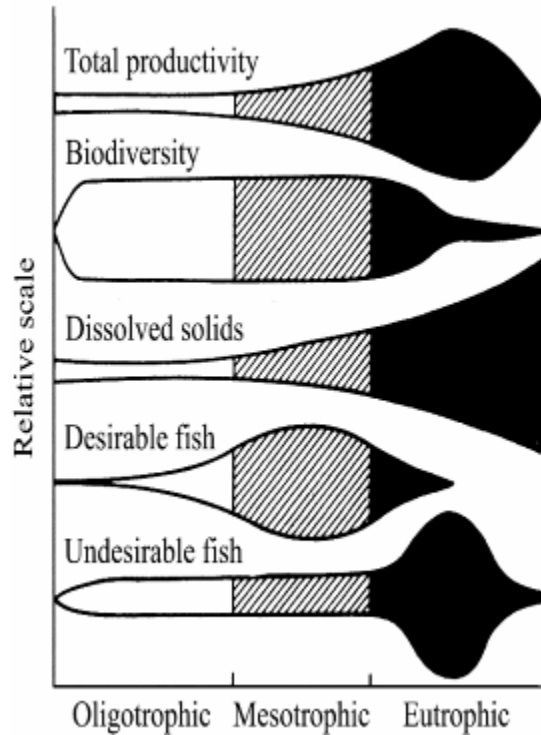


Figure B: Suggested changes in various lake characteristics with eutrophication. From “Ecological Effects of Wastewater”, 1980.

Table A - Trophic status classification based on lake water characteristics.

Trophic state	Total Phosphorus (µg•L <sup>-1</sup> )	Total Nitrogen (µg•L <sup>-1</sup> )	Chlorophyll a (µg•L <sup>-1</sup> )	Secchi Depth (m)
Oligotrophic	< 10	< 350	< 3.5	> 4
Mesotrophic	10 – 30	350 - 650	3.5 - 9	4 - 2
Eutrophic	30 – 100	650 - 1200	9 - 25	2 - 1
Hypereutrophic	> 100	> 1200	> 25	< 1